

# Supporting Information

## Fluorination effects on the thermodynamic, thermophysical and surface properties of ionic liquids

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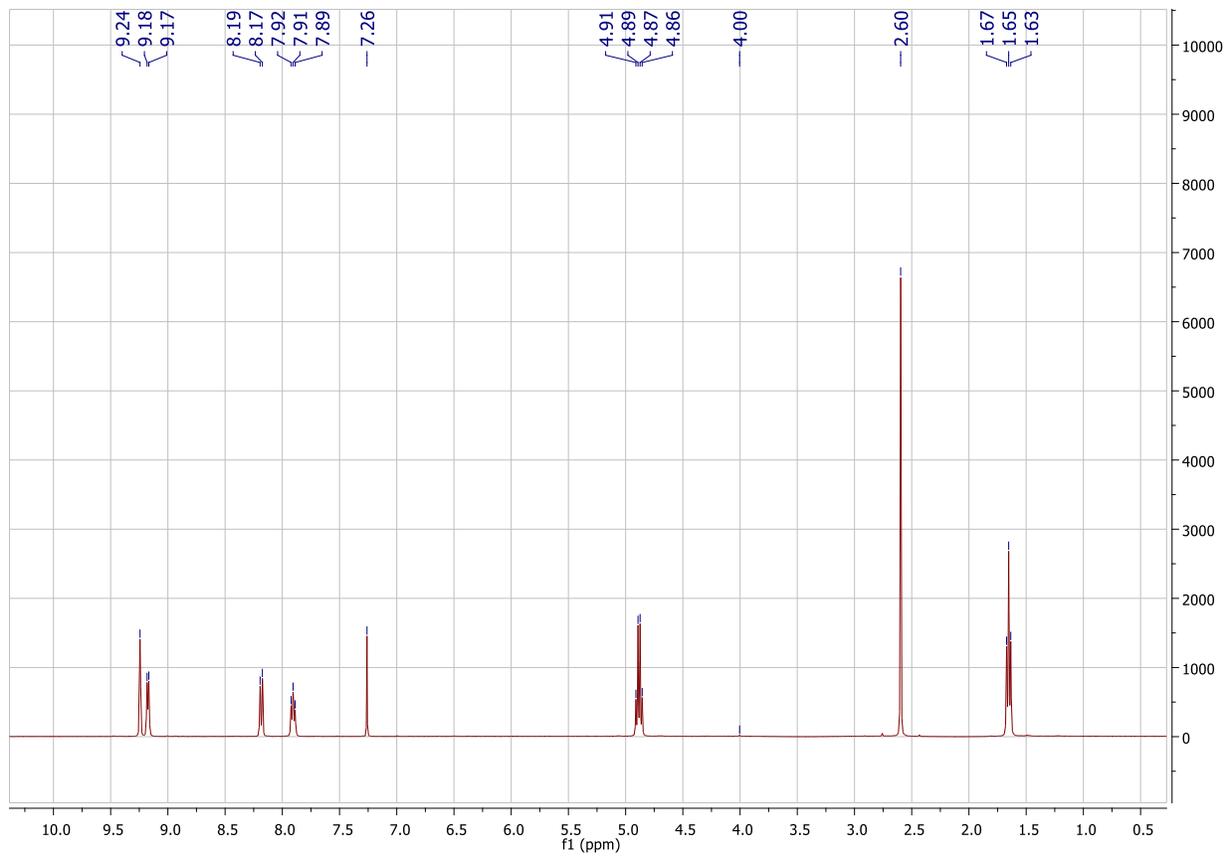
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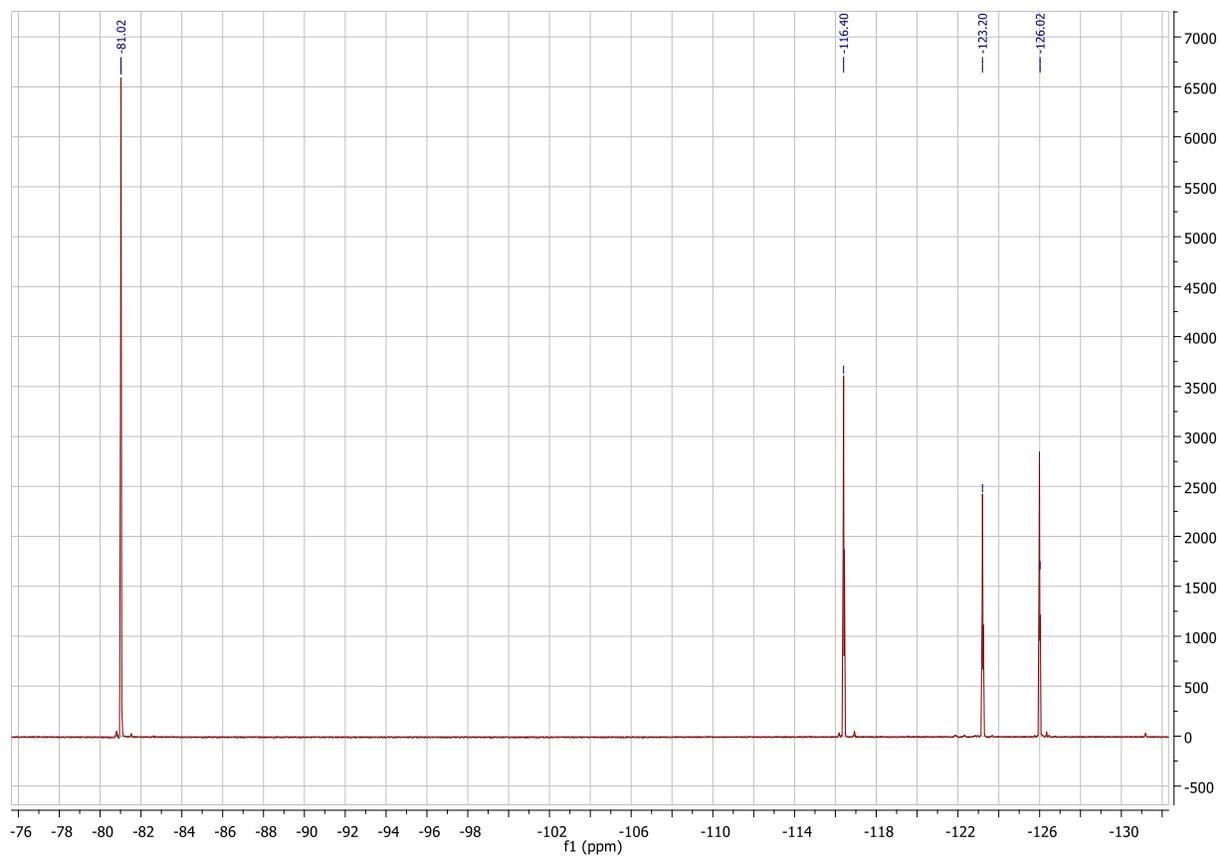
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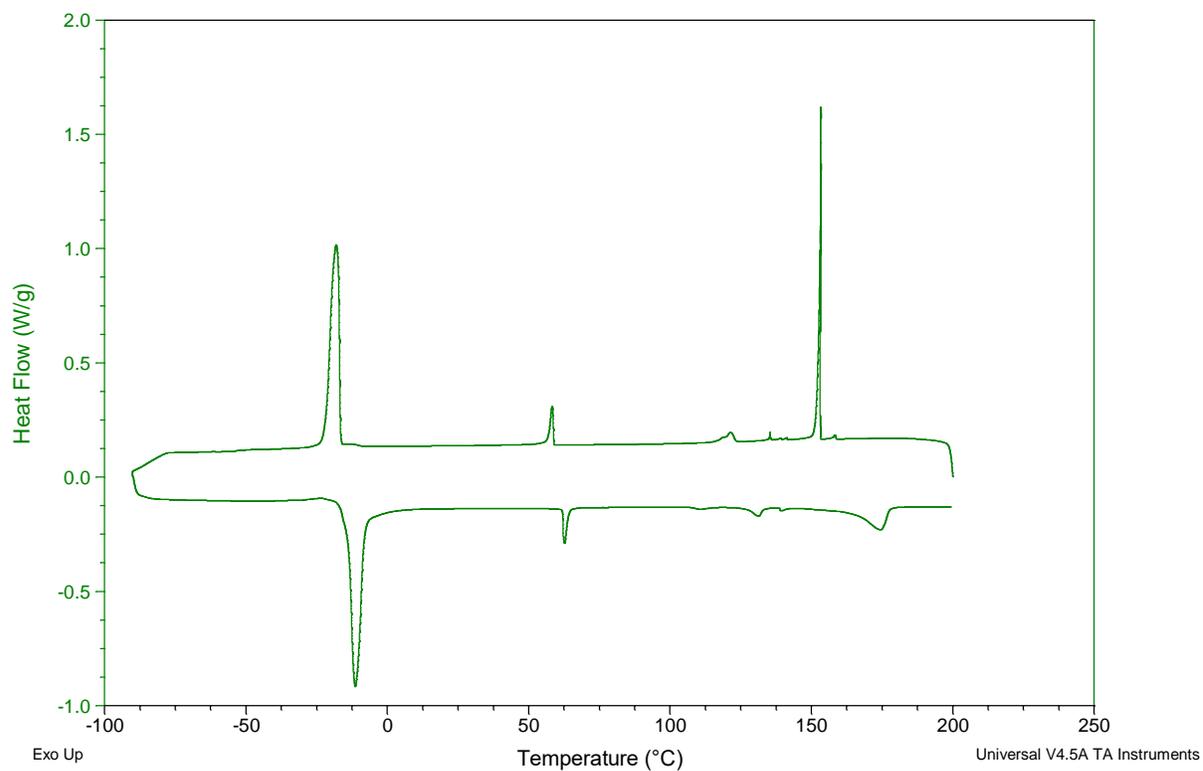
**FIGURE S1.** <sup>1</sup>H NMR spectra of 1-ethyl-3-methylpyridinium perfluoropentanoate, [C<sub>2</sub>C<sub>1</sub>py][C<sub>4</sub>F<sub>9</sub>CO<sub>2</sub>].

<sup>1</sup>H NMR (400 MHz, (CDCl<sub>3</sub>): δ 9.24 (s, 1H, py); 9.18 (d, 1H, py); 8.18 (d, 1H, py); 7.91 (t, 1H, py); 4.88 (q, 2H, CH<sub>3</sub>CH<sub>2</sub>N); 2.60 (s, 3H, CH<sub>3</sub>py); 1.65 (t, 3H, CH<sub>3</sub>CH<sub>2</sub>N).

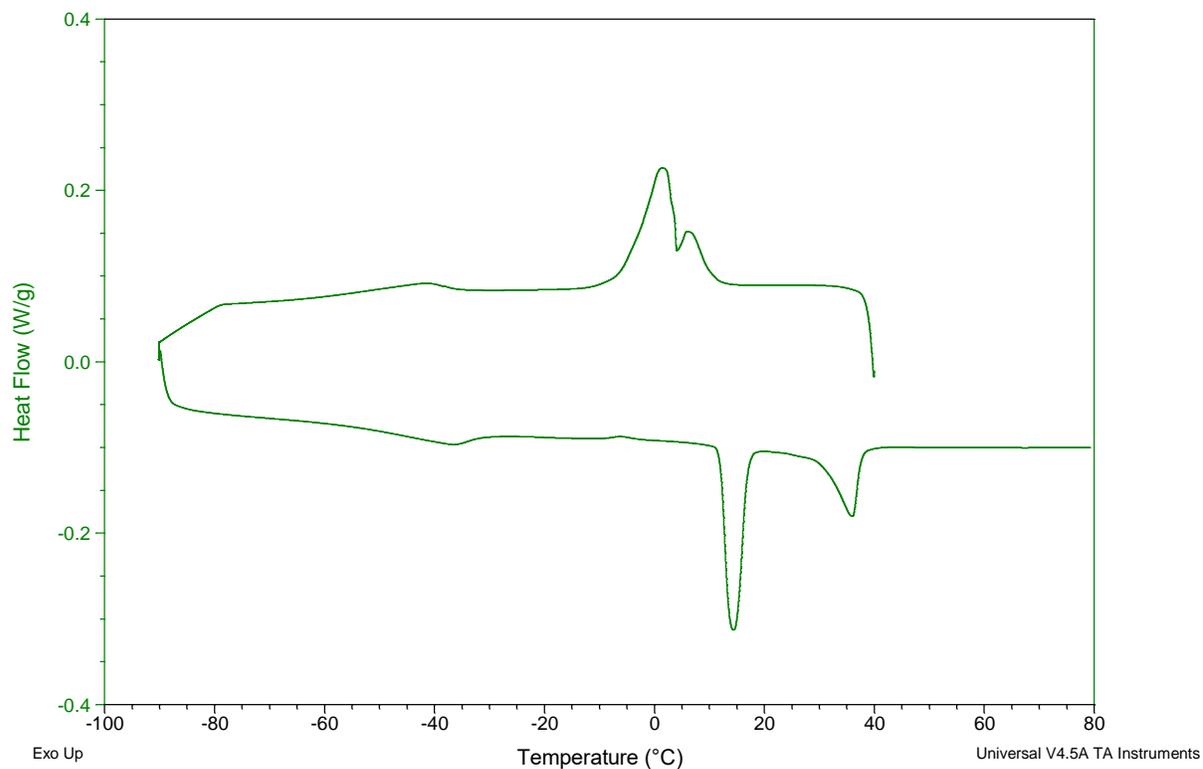


**FIGURE S2.**  $^{19}\text{F}$  NMR spectra of 1-ethyl-3-methylpyridinium perfluoropentanoate,  $[\text{C}_2\text{C}_1\text{py}][\text{C}_4\text{F}_9\text{CO}_2]$ .

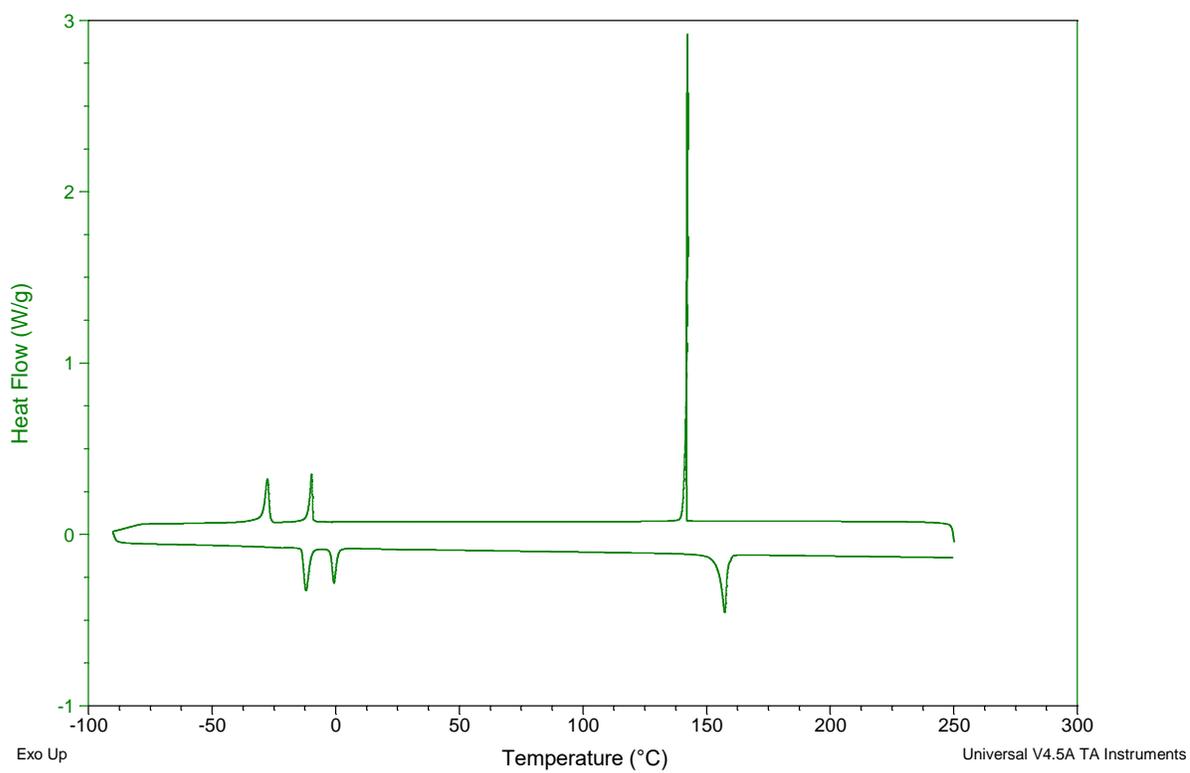
$^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ ):  $\delta$  -81.02 ( $\text{CF}_3$ ); -116.40 ( $\text{CF}_2\text{CO}_2$ ); -123.20 ( $\text{CF}_3\text{CF}_2\text{CF}_2$ ); -126.02 ( $\text{CF}_3\text{CF}_2$ )



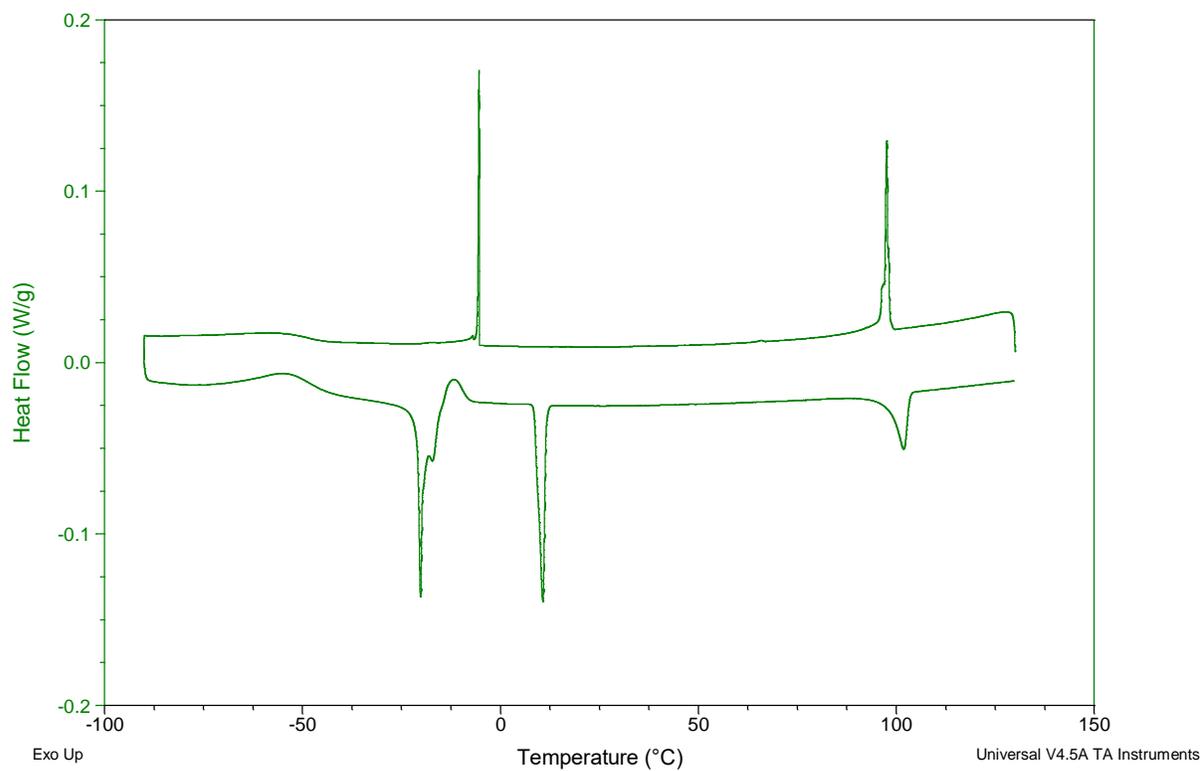
**FIGURE S3.** DSC curve of cholinium perfluorobutanesulfonate,  $[N_{1112(OH)}][C_4F_9SO_3]$ , at a heating rate of 5 K/min.



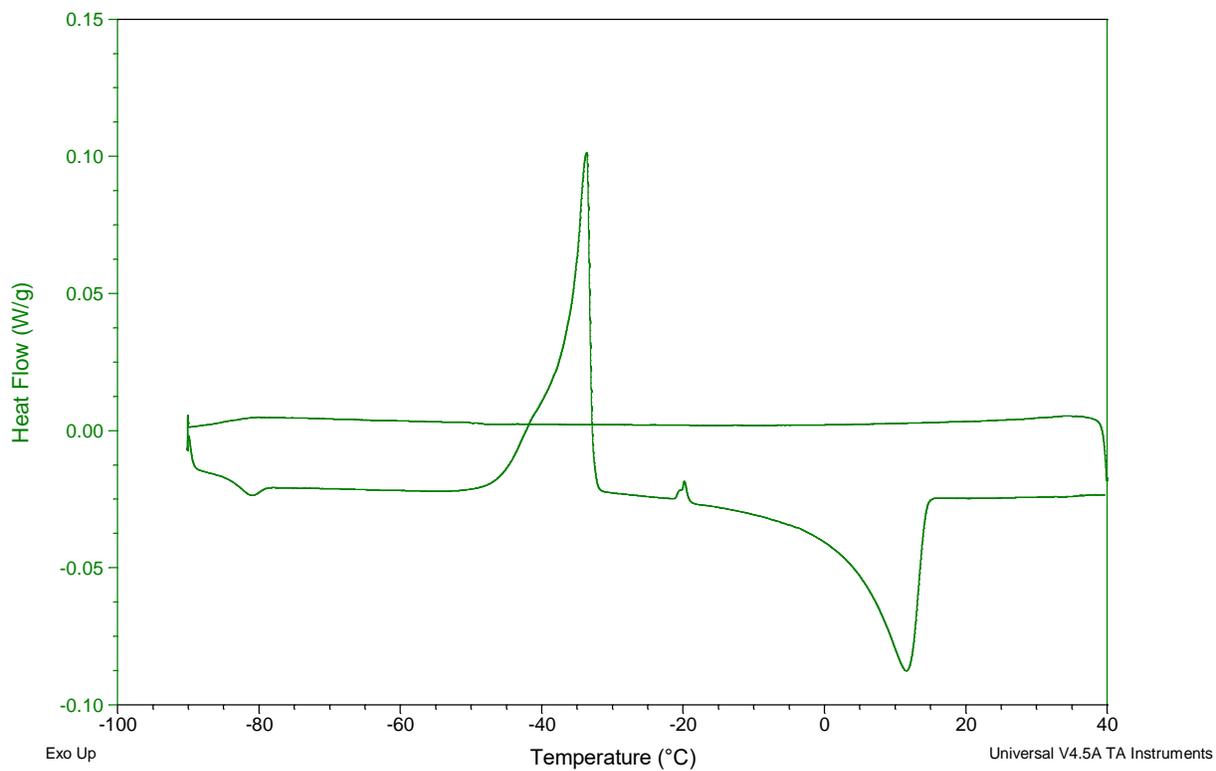
**FIGURE S4.** DSC curve of cholinium bis(nonafluorobutylsulfonyl)imide  $[\text{N}_{1112}(\text{OH})][\text{N}(\text{C}_4\text{F}_9\text{SO}_2)_2]$ , at a heating rate of 5 K/min.



**FIGURE S5.** DSC curve of 1-ethyl-N-methylpyrrolidinium bis(nonafluorobutylsulfonyl)imide[C<sub>2</sub>C<sub>1</sub>pyr][N(C<sub>4</sub>F<sub>9</sub>SO<sub>2</sub>)<sub>2</sub>], at a heating rate of 5 K/min.



**FIGURE S6.** DSC curve of 1-butyl-N-methylpyrrolidinium bis(nonafluorobutylsulfonyl)imide  $[C_4C_1pyr][N(C_4F_9SO_2)_2]$ , at a heating rate of 1 K/min.



**FIGURE S7.** DSC curve of 1-ethyl-3-methylpyridinium perfluoropentanoate,  $[\text{C}_2\text{C}_1\text{py}][\text{C}_4\text{F}_9\text{CO}_2]$ , at a heating rate of 1 K/min.